Chlorine Deactivation in the Lower Stratospheric Polar Regions During Late Winter: Results from UARS

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Short title: CHLORINE DEACTIVATION FROM UARS DATA

The recovery of the perturbed chlorine chemistry in the lower stratospheric polar regions of both hemispheres is investigated using recent data from the Upper Atmosphere Research Satellite (UARS), Microwave Limb Sounder (MLS) measurements of ClO within the polar vortices are correlated with both simultaneous Cryogenic Limb Array Etalon Spectrometer (CLAES) measurements of ClONO₂ and Halogen Occultation Experiment (HALOE) measurements of HCl obtained shortly after the MLS and CLAES data. Time series of vortex-averaged mixing ratios are calculated on two potential temperature surfaces (585 K and 465 K) in the lower stratosphere for approximately month-long intervals during late winter in each hemisphere. The observed mixing ratios are adjusted for the effects of vertical transport using diabatic vertical velocities estimated from CLAES tracer data. In the northern hemisphere, the decrease in ClO is balanced on both surfaces by an increase in ClONO₂. In the southern hemisphere, continuing polar stratospheric cloud activity prevents ClO from undergoing sustained decline until the latter half of the study period. In contrast to the northern hemisphere, there is no significant chemical change in ClONO₂ at 465 K, and there is an apparent decrease in ClONO₂ at 585 K, even after the enhanced ClO abundances have started to recede. Results from an isentropic three-dimensional chemical transport model initialized with UARS data and run with OH + ClO \rightarrow HCl + O₂ as an 8% channel suggest that the primary recovery product in the south during this time period is not ClONO₂, but HCl. HALOE HCl mixing ratios are linearly extrapolated back to the time of the MLS and CLAES data. At 585 K, the chlorine budget can be made to balance by extrapolating back to a value of 0.5 ppbv for the HCl at the beginning of the study period. At 465 K, the contribution from extrapolated HCl is not sufficient to offset the observed loss in ClO, and there is a slight imbalance between the decrease in the reactive chlorine and the change in the chlorine reservoirs. The difficulty in closing the chlorine budget in the southern hemisphere may arise from complications caused by ongoing activation, incomplete photochemical assumptions, or inadequate data quality.

Introduction

Chlorine chemistry is now known to be responsible for the severe depletion in lower stratospheric ozone observed during late winter and early spring over Antarctica [see, e.g., Solomon, 1988, 1990; McElroy and Salawitch, 1989; Anderson et al., 1991; Brune et al., 1991]. Observed decreases in Arctic lower stratospheric ozone have also been shown to be consistent with chlorine catalyzed destruction [e.g., Salawitch et al., 1990, 1993; Schoeberl et al., 1990; Manney et al., 1994a; Traub et al., 1994]. Reactive chlorine is produced through heterogeneous processes occurring on the surfaces of polar stratospheric clouds (PSCs), which form in the low temperatures of polar winter. These heterogeneous reactions rapidly convert the reservoir species HCl and ClONO₂, where most of the inorganic chlorine in the lower stratosphere resides under nonperturbed conditions [Webster et al., 1993], to species such as Cl₂ and HOCl that are easily photolyzed. Following exposure of the processed air to sunlight, the dominant daytime chlorine species is ClO. Maintenance of high ClO levels leads to ozone loss through a catalytic cycle involving ClO dimer (Cl₂O₂) formation [Molina and Molina, 1987].

The timescale for recovery of the perturbed chlorine chemistry in the lower stratosphere is an important topic because the rate at which enhanced ClO abundances recede bears directly on the cumulative ozone loss at high latitudes. One pathway by which reactive chlorine is shifted back to reservoir form is $ClO + NO_2 + M \rightarrow ClONO_2 + M$. In addition to promoting chlorine repartitioning, heterogeneous processes also inhibit reformation of $ClONO_2$ by sequestering reactive nitrogen as HNO_3 in PSC particles. When temperatures rise above the PSC evaporation threshold, gaseous HNO_3 is released; the HNO_3 then photolyzes (on a timescale of ~ 30 days for a solar zenith angle of 70° [Kawa et al., 1992]) to produce NO_2 . If sufficient NO_2 is available, $ClONO_2$ formation takes place in a few hours. Alternatively, chlorine deactivation can proceed via the reaction $Cl + CH_4 \rightarrow HCl + CH_3$. Under non-ozone depleted conditions, the timescale for H^* 1 reformation is roughly comparable to the timescale for HNO_3 photolysis [Kawa et al., 992].

However, [7](21)/ (7 and Jaffe [1990] have Shown that very low ozone concentrations lead to a highly nonlinear chemical system in which the conversion of ClO to HCl is substantially accelerated through an increase in the Cl to ClO abundance ratio. This conversion to 11(Slunder low-ozone conditions has been observed in UARS data by Douglass et al. [1995].

Significant interhemispheric differences in the recovery timescale are expected due t(.) underlying differences in seasonal temperature patterns and vortex behavior [see, e.g., A?Idl'(''Ills et al., 1989; WMO 1991 Report). On average, temperatures in the Arctic vortex are ~15 20 K higher than in the Antarctic, and the Arctic vortex is much less st able, leading to fewer and less persistent PSC events in the Arctic. Denitrification the removal of nitrogen from the lower stratosphere through gravitational settling of PSC particles containing HNO₃ is much less extensive in the Arctic than in the Antarctic [cf. Fahey et al., 1989; Toon et al., 1989; Coffey et al., 1989: Mankin et al., 1990; Kawa et al., 1990; Poole and Pitts, 1994; Santeε et al., 1995]. Although enhanced ClO abundances have been observed in the polar vortices of both hemispheres $[de\ Zafra]$ et d., 1987; Anderson et al., 1989; Waters et al., 1993a, b], the accompanying 10ss of Arctic ozone has been much less severe [Manney et al., 1994 a] than that over Antarctica. The warmer temperatures, greater dynamical activity, and continued presence of $IINO_3$ throughout northern winter moderate ozone loss and prevent the formation of an Arctic ozone "hole" [Brune et al., 1991; Schoeberl and Hartmann, 1991; Santee et al., 1995]. Differences in the late winter/carly spring HNO₃ and O₃ concentrations between the two hemispheres lead to dissimilarities in the production rates of the chlorine reservoirs.

Chlorine deactivation has been the subject of several previous investigations. Toohey et al. [1993] and Webster et al. [1993] report simultaneous in situ measurements of ClO and HCl from the second Airborne Arctic Stratospheric Expedition (AASE II) in the 1991-1992 Arctic winter. They find that decreasing ClO abundances during the recovery phase are consistent with the rapid formation of ClONO₂ (calculated from an estimate of

the total available inorganic chlorine), while HCl remains substantially depleted through the end of the observational period in February 1992. Contemporaneous observational [Adrian et al., 1994; Oelhaf et al., 1994; Bell et al., 1994] and modeling [Lutman et al., 1994; Müller et al., 1994] studies from the European Arctic Stratospheric Ozone Experiment (EASOE) find the same recovery pattern: after cessation of PSC activity, there is an almost complete conversion of active chlorine to ClONO₂, which remains the dominant inorganic chlorine compound for more than a month. These recent findings confirm the conclusions from earlier modeling studies constrained by AASE observations [Jones et al., 1990; McKenna et al., 1990]. In Antarctica, infrared solar spectra recorded at McMurdo in late winter/early spring 1986 indicate a twofold increase in column HCl + ClONO₂ associated with rapidly rising temperatures [Farmer et al., 1987]. While the sum of HCl and ClONO_2 remains roughly constant thereafter, the $\text{ClONO}_2/\text{HCl}$ ratio gradually declines from 1.5 in late September to 0.3 in late October. Liu et al. [1992] also use HCl column amounts retrieved from infrared solar spectra obtained at McMurdo, in combination with a one dimensional photochemical model, to show that almost all active chlorine species are converted back to HCl by mid-October. Rapid conversion of ClO to the reservoirs ClONO₂ and HCl occurs once PSCs evaporate, followed by slow repartitioning between the reservoirs. Schoeberl et al. [1993] perform trajectory analyses of ClO and NO measurements from both Arctic and Antarctic aircraft campaigns and conclude that production of ClONO₂ is the main mechanism for the initial decay of high ClO concentrations when the air is not denitrified; in highly denitrified parcels, lack of HNO₃ (and therefore NO₂) prevents transformation of ClO to ClONO₂, and ClO remains enhanced.

In this paper, we use recent data from the Upper Atmosphere Research Satellite (UARS) to examine chlorine reservoir recovery in the lower stratospheric polar regions of both hemispheres. Microwave Limb Sounder (MLS) measurements of ClO within the polar vortices are correlated with both simultaneous Cryogenic Limb Array Etalon

Spectrometer (CLAES) observations of (; IONO₂ and Halogen Occultation Experiment (HALOE) observations of HClobtained shortly after the M 1,S and (LAES data. We analyze time series of vortex- averaged mixing ratios on two potential temperature surfaces (585 K and 465 K) in the lower stratosphere for approximately month-long intervals during late winter: August 17 - September 17, 1992 in the southern hemisphere and February 12 March 16, 1993 in the northern hemisphere. We focus on the late winter balance between the decay of reactive chlorine and the growth of chlorine reservoir species based on these data sets, and we compare the observations] results with those from a three-dimensional chemical transport model [see the companion paper by Chipperfield et al., 1995] initialized with UARS data.

Data and Instrument Description

MLS acquires stratospheric measurements of millimeter wavelength thermal emsion that are not degraded by PSCs or aerosol. The measurement technique and the instrument are described in detail by Waters [1993] and Barath et al. [1993], respectively. MLS observations of ClO have been presented previously by Waters et al. [1993a,b; 1995a] and Manney et al. [1994a]. The estimated precision (rms) and absolute accuracy of an individual ClO profile in the lower stratosphere are ~0.5 ppbv (parts per 10⁹ by volume) and 15–20%, respectively [Waters et al., 1995b] for Version 3 files. In this study, we utilize ClO values from preliminary algorithms that also retrieve HNO₃, eliminating a bias error in the Version 3 ClO data [Waters et al., 1995b]. These ClO data have an accuracy of about 10%.

CLAES [Roche et al., 1993a] measures infrared thermal emission in nine spectral regions between 3.5 μ m and 13 μ m. MLS and CLAES observations are essentially simultaneous and co-located (typically to within a few tens of kilometers). CLAES ClONO₂ measurements have been reported by Roche et al. [1993b; 1994]; here we use Version 7 files, which have an estimated systematic error of less than 20% and a single-profile pre-

cision of ~0.2 ppbv in the lower stratosphere [Mergenthaler et al., 1995]. To eliminate spikes and other known artifacts in the data, all points for which the ClONO₂ value either exceeds 3.5 ppbv or exceeds 1.5 ppbv with an associated error of greater than 50% are excluded from the analysis [Mergenthaler et al., 1995]. We also utilize CLAES observations of the dynamical tracers CH₄ and N₂O [Kumer et al., 1993; Roche et al., 1995]. For CH₄, the estimated systematic error and precision are about 15% and 0.1 ppmv, respectively, at 46 hPa, and the mixing ratios are most reliable for values below 1.5 ppmv. For N₂O, the estimated systematic error and precision are about 20% and 20 ppbv, respectively, at 46 hPa, and the mixing ratios are most reliable for values below 250 ppbv. In addition to constituent mixing ratios, aerosol extinction coefficients are also retrieved from the measured spectra [Mergenthaler et al., 1993; Massic et ale, 1995]. The aerosol data we show here are from the 790 cm⁻¹ spectral region, where there is strong spectral contrast between gas (O₃ and CO₂) line emission and aerosol continuum emission.

Both CLAES and MLS are situated on the anti- sun side of the spacecraft, and both of their optical pointing axes are directed approximately 23° down from the local horizontal at the observation point. This pointing geometry, coupled with the 57° inclination of the UARS orbit, leads to measurement coverage from 80° latitude on one side of the equator to 34° on the other. The UARS orbit plane precesses in such a way that all local solar times are sampled in about 36 days (getting ~20 min earlier each day at a given latitude), after which the spacecraft is rotated 180° about its yaw axis. Thus, 10 times per year MLS and CLAES alternate between viewing northern and southern high latitudes.

HALOE uses solar occultation to measure the attenuation of infrared solar energy due to stratospheric constituents [Russell et al., 1993]. In this study, we use Version 16 HCl data, which are described by Russell et al. [1995]. Because measurements are made only at the spacecraft sunrise and sunset times, daily sampling typically consists of 15 sunrise profiles at one particular latitude, and 15 sunset profiles in the opposite

hemisphere. The 11 ALOE observing pattern moves slowly north and south throughout the year, with wintertime coverage of the most extreme southern latitudes (nearly to 80°S) occurring from mid-September through mid-October, and wintertime coverage of the most extreme northern latitudes (nearly to SOON) occurring from mid-March through mid-April. In the 1992 southern and 1992-93 northern late winter periods, HALOE measurements are not obtained within the polar vortices until after the spacecraft has yawed towards the opposite hemisphere; thus, the HALOE high-latitude data are not coincident with either the MLS or CLAES data.

1992 Southern Hemisphere Winter

Behavior of ClO and ClONO₂

As discussed in the Introduction, high levels of CIO in the vortex follow from the heterogeneous activation of chlorine as temperatures fall below the PSC formation threshold (~195 K [Turco et al., 1989]) in early winter. Elevated levels (mixing ratios greater than 1 ppbv) of lower stratospheric ClO were observed in portions of the southern polar vortex as early as 2 June (the beginning of an MLS south-viewing period) [Waters et al., 1993a,b]. ClO abundances continued to increase throughout the winter, and at the start of the next south-viewing period on 17 August 1992, MLS measured lower stratospheric ClO values of more than 2 ppbv in most of the sunlit area poleward of 60°S [Waters et al., 1993a,b]. It is for this late winter (17 August - 17 September, 1992) yaw period that we will examine the relationship between the decrease in ClO and the increase in the reservoir species ClONO₂ and HCl.

The behavior of MLS CIO on three selected days at the beginning, middle, and end of the southern hemisphere late winter yaw period is shown in Figure 1a. The data have been interpolated onto constant potential temperature (Θ) surfaces at 585 K (corresponding to \sim 24 km, 20 hPa for the cold temperatures characteristic of the polar

vortex) and 465 K (~19 km, 50 hPa) using United States National Meteorological Center (NMC) temperatures. Only data from the "day" side of the orbit are shown because ClO abundances drop sharply in regions where the solar zenith angle (Figure 1b) is greater than $\sim 94^{\circ}$, due to lack of Cl_2O_2 photolysis. Superimposed on each of the maps are three contours of potential vorticity (PV), calculated from NMC temperatures and derived winds using the algorithm described by Manney and Zurek [1993]. These PV contours are used to delimit different areas over which averaged mixing ratios are calculated. The outermost contour represents a typical definition of the vortex edge during winter [e.g., Manney et al., 1993] and coincides with a strong barrier to mixing [Manney et al., 1994b]. The innermost contour on each Θ surface has a PV value twice that of the outermost one. The intermediate contour is situated poleward of the vortex boundary, and using it reduces the potential impact of horizontal transport from lower latitudes on the averaged mixing ratios. It more closely delineates the region of high ClO at 465 K in mid-August, and it also closely approximates the edge of the region of low values of N_2O and CH_4 (not shown) caused by stronger diabatic descent within the vortex. The PV contours indicate that the vortex is approximately symmetric with respect to the pole throughout this period, and its size, shape, and strength are roughly constant at both levels.

From mid-August to mid-September, the outside edge of enhanced CIO retreats poleward. Elevated CIO abundances recede more rapidly at 585 K than at 465 K, with the majority of the decrease occurring in the latter half of the period. The area poleward of 52°S in which daytime CIO mixing ratios exceed 1 ppbv, shown in Figure 2, is a diagnostic of the time and altitude dependence of the decay of activated chlorine. The value of 52° is selected to generally encompass the maximum areal extent of the vortex at these levels during this time period while excluding lower latitudes (where CIO is not enhanced). The lower stratospheric minimum values occurring at the beginning of September are artifacts caused by a brief interval in the middle of the yaw period during which there are essentially no daylight measurements due to precession of the orbit. The

slope and spacing of the contours in Figure 2 illustrate the more rapid decline in ClO at the higher Θ levels, and that this decline intensifies at the beginning of September.

In the southern polar region, PSCs typically persist at 18 km until at least midto late September [McCormick et al., 1989; Poole and Pitts, 1994]. In 1987, PSCs were present at this level into mid-October [McCormick et al., 1989; Poole and Pitts, 1994], and results from ER-2 aircraft flights between 23 August and 22 September 1987 show that ClO mixing ratios at 18 km increased throughout this period [Brune et al., 1989]. Jones et al. [1989] use the ER-2 observations from early September 1987 to initialize a photochemical model integrated along air parcel trajectories, and find that maintenance of elevated ClO concentrations is very dependent on the frequency and duration of PSC events. In this study, we use CLAES measurements of aerosol extinction to assess the possibility of continued chlorine activation during our observational period. The area in which aerosol extinction coefficients in the region poleward of 52°S exceed 1.0 $\times 10^{-3}$ km⁻¹ [Turco et al., 1989] is given in Figure 3. At the beginning of the late winter yaw period in mid-August, PSC activity is as extensive as it was during the previous June-July south-looking interval (not shown), and heterogeneous chlorine activation is expected to be ongoing. Maps of aerosol extinction on individual days (also not shown) indicate that, although there are no significant PSC events observed to extend up to the 585 K level after the first few days of September, there continue to be intermittent localized PSC events at 465 K as late as 15 September. These episodic PSCs may partially explain the slower decline in ClO at 465 K than at 585 K. Because the focus here is on the deactivation of chlorine, we will limit the study to the latter half of the yaw period (starting on 3 September), when PSC activity is beginning to subside and (as seen in Figure 2) the high ClO abundances are receding.

If the primary reservoir for chlorine during the initial stages of recovery were ClONO₂, then production of ClONO₂ would parallel the reduction in ClO evident in Figures 1 and 2. Maps of CLAES ClONO₂ during the late winter interval are shown in Figure 4. To ensure that the same air parcels are being considered for both (3O and ClONO₂, these maps are also constructed using data from the "day" side of the orbitonly. As discussed by Roche et al. [1 9936; 1994], ClONO₂ mixing ratios rise steeply from mid-latitudes to form a roughly circumpolar collar of enhanced abundances, with some intrusion of high ClONO₂ into the vortex core, where values are generally much smaller than in the collar region. By mid-September, the high values of the collar region have diminished at both Θ levels, and an overall increase in ClONO₂ comparable to the decrease in ClO is not seen. As shown in Figure 5, the area poleward of 52°S distinguished by high ClONO₂ abundances (> 1.5 ppbv) is essentially constant at 465 K but decreases at 585 K throughout the yaw period.

Effects of Vertical Transport

One of the issues that must be addressed in contrasting the behavior of ClO and ClONO₂ is the possible effect of vertical transport on the mixing ratios at these levels. As the polar regions fall into darkness after the autumnal equinox, radiative processes rapidly cool the polar stratosphere, the vortex spins up, and strong downward motion over this area begins [Schoeberl and Hartmann, 1991]. The descending air carries with it the chemical composition of the upper stratosphere, and therefore, depending on the vertical profile of a given species, can produce steep horizontal gradients in its concentration on isentropic surfaces. Both N₂O and CH₄ have long chemical lifetimes with tropospheric sources and stratospheric sinks, and have been shown to be useful as conservative tracers of stratospheric air motions [Locwenstein et al., 1989, 1990; Collins et al., 1993].

We estimate the vertical velocities at 585 K and 465 K from CLAES measurements of N_2O and Cll_4 . Vortex-averaged mixing ratios are calculated by dividing the area integral of the mixing ratio inside a PV contour by the area enclosed by that contour. This calculation is performed for the three PV contours shown in Figure 1a. Similar results are obtained for all PV values; here we discuss the results from the intermediate

contour only. Daily vortex- averaged mixing ratios for both N₂O and CH₄ are shown in Figure 6 as a function of time. For perfect tracers in the absence of diabatic descent, the lines fit to the averaged mixing ratios would have zero stope. Under the assumption that the trends exhibited by these species arise solely from diabatic descent, the slopes of these lines are used in the tracer continuity equation to estimate average vertical velocities., The vertical velocities estimated from the two tracers are generally consistent, and are averaged together to obtain the overall estimate of tile vertical velocity at Cach (-) level. Uncertainties are assigned by propagating into the velocity calculation the uncertainties in the slopes of the lines fit to the daily vortex-averaged mixing ratios and the standard deviations of the tracer vertical profiles, and then augmenting the resulting error if necessary to ensure that the overall average encompasses the estimates from both tracers. This is a deliberately conservative process intended to provide an upper limit on the uncertainty estimate.

Based 011 this approach, we find the vortex-averaged descent rate to be 0.5±0.3 illill/S at 585 K and 0.7 ± 0.'2 mm/s at 465 1{ when only the latter half of the yaw period is considered (velocities are higher when trends are taken over the entire yaw period, particularly at 585 K). It should be noted that these are vortex-averaged values; locally, velocities could depart significantly from the vortex averages. Within the uncertainties, these values are in agreement with estimates of the lower stratospheric vertical velocity (1.0) mm/s) in August 1992 obtained from simulated air parcel trajectories by Manney et al. [1994 6]. They also agree very well with the average lower stratospheric descent rates (1.5-1.8 km/month, or 0.6-0.7 mm/s) estimated for tile 1992 winter season from UARS IIALOE CII₄ trends [Schoeberl et d., 1995]. However, they are larger than the late winter residual vertical velocities diagnosed from 1987 NM(; temperatures and Airborne Antarctic Ozone Experiment (AAOE) 11₂0 and O₃ data by Schoeberl et al. [1 992], which range from 0.1to 0.2 111111/s between 18 and 22 km at 70°S. More recent radiative transfer calculations by Rosenfield et al. [1 994], based 011 NMC temperatures during both the

1987 and 1992 Antarctic winter seasons, also produce smaller average diabatic descent rates in the lower stratospheric vortex (~0.4-0.9 km/month, or ~0.2-0.3 mm/s) than those estimated here from the CLAES tracer data. It is important to note, however, that the rates deduced by *Rosenfield et al.* [1994] represent an average computed from the descent of air over an eight-month period from March through October, and a direct comparison with our late-winter velocity estimates is an oversimplification.

Once the descent rate at each level has been estimated, the changes in ClO and ClONO₂ due to vertical transport can also be calculated from the tracer continuity equation, assuming constant velocities throughout the two-week interval. This calculation would not be necessary for species that are short-lived compared to a two-week interval, because local photochemical conditions -- not dynamics -- drive changes in their abundances. However, total reactive chlorine ClO_x ($ClO_x = ClO + 2Cl_2O_2$) is affected by dynamics, and it in turn affects ClO and ClONO₂. As discussed further below, the approximate nature of the approach used here to account for the effects of dynamics is not a major factor in the overall conclusions. The impact of diabatic descent on the mixing ratios of a constituent depends not only on the magnitude of the vertical velocity but also on the shape of the constituent's vertical profile. The ClO mixing ratio profile (Figure 7a) drops sharply above 465 K, and the downward transport of air with lower ClO abundances accounts for a portion of the observed decline in ClO values at both 585 K and 465 K. Because the peak of the ClONO₂ profile (Figure 7b) occurs near 585 K, descent is estimated to produce a decrease in ClONO₂ concentrations at that level, but an increase in ClONO₂ concentrations at 465 K. Figure 8 shows daily vortex averages of the observed ClO and ClONO₂ abundances, along with adjusted values obtained by subtracting the estimated changes in mixing ratio due to diabatic descent. These adjusted mixing ratio values reflect the trends in the species that can be attributed to chemical processes. The results presented in Figure 8 are based on averages within the intermediate PVcontour in Figure 1a, restricted to the latter half of the yaw period (after 3 September). Similar behavior is found for vortex averages within the other two PV contours. For this period, the implied chemical loss in ClO is approximately 0.4 ± 0.1 ppbv at 585 K and 0.:130.1 ppbv at 465 K, based on linear fits to the daily averages. Although there is a clear decreasing trendin ClO throughout this interval at both levels, there is an apparent decrease of 0.1 ± 0.1 ppbv in the adjusted $ClONO_2$ mixing ratios at 585 K, and there is no significant chemical change in $ClONO_2$ at 465 K ($\sim 0.0 \pm 0.1$ ppbv). No increase in $ClONO_2$ is found to offset the decrease in ClO even when the values at the extreme limits of the uncertainties are used. These adjusted ClO and (; $lONO_2$ concentrations are computed using the nominal magnitudes of the vertical velocities; however, calculations over the range of velocity values defined by the uncertainties yield very similar results. In fact, sensitivity tests indicate that for there to be an increase in $ClONO_2$ balancing the decrease in ClO there would have to be very strong descent (~ 2.3 mm/s) at 585 K but ascent (~ 0.7 mm/s) at 465 K.

To ensure that a consistent set of air parcels is being evaluated for all species, in the preceding calculations all vortex averages include data from the "day" side of the orbit only. However, if any of the chlorine released from ClO is being converted into ClONO₂, then ClONO₂ mixing ratios at night could be slightly larger than those during the day, when ClONO₂ is subject to photodissociation. We therefore examine vortex averages of ClONO₂ mixing ratios from the "night" side of the orbit, adjusted using the diab atic vertical velocities calculated above. in this case there is no change ($\sim 0.0 \pm 0.1 \,\mathrm{ppbv}$) at $5.85\,\mathrm{K}$, and a decrease of $0.1 \pm 0.1 \,\mathrm{ppbv}$ at $465\,\mathrm{K}$. As before, no increase in ClONO₂ is found to offset the decrease in ClO even when the values at the extreme limits of the uncertainties are used.

Results from the Chemical Transport Model

We also use the SLIMCAT isentropic three dimensional chemical transport model [see the companion paper by Chipperfield et al., 1995] to investigate the decay of enhanced

1987 and 1992 Antarctic winter seasons, also produce smaller average diabatic descent rates in the lower stratospheric vortex (~ 0.4 –0.9 km/month, or ~ 0.2 -0.3 mm/s) than those estimated here from the CLAES tracer data. It is important to note, however, that the rates deduced by Rosenfield et al. [1 994] represent an average computed from the descent of air over an eight-month period from March through October, and a direct comparison with our late-winter velocity estimates is an oversimplification.

Once the descent rate at each level has been estimated, the changes in (30 and (IONO₂ due to vertical transport can also be calculated from the tracer continuity equation, assuming constant velocities throughout the two-week interval. This calculation would not be necessary for species that are short-lived compared to a two-week interval, because local photochemical conditions -- not dynamics - drive changes in their abundances. However, total reactive chlorine ClO_x ($ClO_x = ClO + 2Cl_2O_2$) is affected by dynamics, and $\,$ it $\,$ in turn affec ts ClO and ClONO $_2$. As discussed further below, the approximate nature. Of the approach used here to account for the effects of dynamics is not a major factor in the overall conclusions. The impact of diabatic descent on the mixing ratios of a constituent depends not only on the magnitude of the vertical velocity but also on the shape of the constituent's vertical profile. The ClO mixing ratio profile (Figure 7a) drops sharply above 465 K, and the downward transport of air with lower (!IO abundances accounts for a portion of the observed decline in CIO values at both 585 K and 465 K. Because the peak of the ClONO₂ profile (Figure 7b) occurs near 585 K, descent is estimated to produce a decrease in $ClONO_2$ concentrations at that level, but an increase in (!IONO₂ concentrations at 465 K. Figure 8 shows daily vortex averages of the observed (30 and (30NO₂ abundances, along with adjusted values obtained by subtracting the estimated changes in mixing ratio due to diabatic descent. These adjusted mixing ratio values reflect the trends in the spec less that can be attributed to chemical processes, The results presented in Figure 8 are b ased on averages within the intermediate PVcontour in Figure 1a, restricted to the latter half of the yaw period (after 3 September). CIO in the Antarctic vortex. ECMWF meteorological analyses are used to specify the horizontal winds and temperatures with a spectral truncation of T42, yielding a model horizontal resolution of 2.8° x 2.8°. The model employs a detailed stratospheric chemistry scheme with treatment of heterogeneous reactions on PSCs and sulfate aerosols. It is initialized with UARS data for 31 August 1992 interpolated onto the '1'42 Gaussian grid and is run on both the 585 K and 465 K surfaces. In the companion paper, Chipperfield et al. [1995] report on results from the basic model as well as those from several sensitivity tests; here we present results only from run I) of the companion paper. This run treats 465 K and 585 K as separate unconnected isentropic surfaces with no vertical transport between them. It also includes the reaction OH+ClO -> HCl+O₂ as an 8% channel. This pathway has been shown to improve the agreement, between photochemical models and observations of ClO,HCl and 03 in the upper stratosphere [McElroy and Salawitch,1989; Natarajan and Callis, 1991; Chandra et al., 1993; Toumi and Bekki,1993; Minschwaner et al., 1993]. Chipperfield et al. [199(5] show that the inclusion of this channel is also important in the lower stlal,c~sl}lll['if:, affecting model results at both 585 K and 465 K.

Vortex averages of model chlorine species from this run are shown in Figure 9, along with MI, S CIO and CLAES CIONO₂ data. The model fields are sampled at 1200UT except for CIO and CIO NO₂, which are sampled at the same local time as the MLS and CLAES measurements. Total inorganic chlorine (ClO_y) is essentially constant at 3.0 ppbv at 585 K and 2.8 ppbv at 465 K. The model reproduces the M LS CIO abundances and downward trend at 585 K, but at 465 K it underestimates MI,S (ClO by up to 0.4 ppbv (see Chipperfield et al. [1995] for an elaboration of the low model ClO). At 465 K it also indicates a slightly increasing trend in ClO (not seen in the data) which is probably related to the fact that the model overestimates the extent of the ongoing PSC processing. Similarly, the model reproduces the CLAES ClONO₂ data fairly well at 585 K (although it indicates a small increase in ClONO₂ over the interval which is not seen in the data), but at 465 K it underestimates CLAES (ClONO₂ by as much as 0.6 ppbv. As discussed

by Chipperfield et al. [1 995], inclusion of the OH+ClO→HCl+O₂ reaction suppresses the increase in ClONO₂ and promotes the conversion of ClOtoHCl. The calculated changes in ClONO₂ and HOClare not significant, and the main recovery is predicted to be into HClat both 585 K and 465 K.

For the calculations presented in Figure 8 we have neglected the effects of the dimer, ${\rm Cl_2O_2}, {\rm produced\ from\ ClO\,recombination}\colon {\rm ClO} + {\rm ClO} + M \to {\rm Cl_2O_2} + M.$ Since there are no measurements of Cl₂O₂, its contribution is assessed from model results. The simulations show a more rapid decay in ClO_x than in ClO, particularly at 585 K (Figure 9). That is, as the ClO_x diminishes throughout the study period, the partitioning between ClO and Cl₂O₂ increasingly favors ClO and the apparent decrease in ClO is reduced. Had ClO_x been used in place of ClO in the foregoing analysis, the lack of balance between the decrease in reactive chlorine and the increase in ClONO₂ would have been amplified. Another factor which could impact the observed rate of ClO decline is the changing solar zenith angle and local solar time of the measurements. The dimer is cycled back into ClO through the processes of thermal decomposition and photolysis, which is curtailed under low-sunlight conditions. The solar zenith angle at 75°S decreases from ~88° to \sim 78° in the period after 3 September (cf. Figure 1b), leading to increased ClO relative to ClO_x . '1'0 address this issue, model results from run D are sampled at local noon (when ClO is expected to be a maximum) at every grid point and compared to those from sampling at the same local time as the measurements (not shown; see Chipperfield et al. [1995]). In early September, there is a slight (~ 0.1 ppbv) difference at both levels in vortex averaged model ClO between the two sampling strategies which decreases throughout the interval as the local solar time of the MLS measurements in the vortex region shifts from mid-afternoon to late morning. In general, the change in the solar zenith angle has only a small influence on the ClO concentrations (at the solar zenith angles in this study period), and we conclude that it can be safely ignored in analyzing the observations.

Behavior of HCl

As the model results of Chipperfield et al. [1995] suggest, CIONO₂ may not be the dominant chlorine reservoir during the initial stages of recovery. Santee et al. [1995] report that in mid-August 1992, gas-phase HNO₃ values measured by MLS were extremely low (≤ 2 ppbv) throughout a low-temperature (≤ 195 K) region encompassing most of the Antarctic vortex. The deficit in gas-phase HNO₃ persisted into November, well after lower stratospheric temperatures had risen above the PSC evaporation threshold, implying that denitrification had occurred in the polar vortex. Removal of gas-phase HNO₃ prevents NO₂ release from HNO₃ photolysis, and consequently inhibits formation of ClONO₂. Detailed modeling simulations [Prather and Jaffe, 1990] of perturbed air parcels spun out from the vortex and transported to lower latitudes have shown that when denitrification occurs, ClONO₂ is a less significant chlorine reservoir during the first 20 days, and recovery is limited by formation of HCl. Therefore, we consider the possibility that the reactive chlorine is being converted into HCl rather than ClONO₂, as seen by Douglass et al. [1995].

HALOE observations of HCl become available within the southern polar vortex on about 23 September, a few days after the emission instruments turn away from the south and resume viewing northern high latitudes. HALOE sampling of the southern polar region ceases around 27 October. Individual sunrise or sunset profiles are interpolated to Θ surfaces, and a simple average of all of the profiles falling within the intermediate PV contour shown in Figure 1a is calculated. The number of points comprising such a "vortex average" varies with the latitude of the profiles: the maximum number is 15. The daily vortex- averaged mixing ratios at 585 K and 465 K are presented in Figure 10 for the entire period of southern vortex coverage, along with their associated daily average latitudes. Vortex-averaged HCl abundances are seen to increase during this period, particularly at 465 K. Vortex averages of model HCl (which is calculated out through the beginning of October; not shown) agree very well with those of HALOE HCl at

465 K, but underestimate those of HALOE HCl at 585 K by about ().3 ppbv.However, synoptic maps of HCl from the model (Figure 9 of Chipperfield et al. [1995]) indicate that the recovery of HCl in the polar region is not homogeneous - at 585 K there is a maximum in model HCl along the vortex edge, with values decreasing toward the vortex center, whereas at 465 1{ the opposite behavior is predicted. Therefore, since the vortex averages in Figures 1 0a and 10b are based on data from one particular latitude on any given day (see Figure 10c), they may not be representative of the vortex HCl as a whole.

Because the HALOE data are strong functions of latitude as well as time, multiple linear regression (with both latitude and time functions included) is used to fit a curve to the daily averages as in $Schoeberl\,et\,al.$ [1–995]. We initially use the $\partial(HCl)/\partial t$ value obtained from this fit to extrapolate the HCl values back to the timeframe of the MLS and CLAES data. However, at 465 K the change in HCl with time is sufficiently rapid that for most of the interval the extrapolation produces negative mixing ratios. In contrast, at 585 K the change in HCl with timr is so gradual that at the beginning of the interval the extrapolation produces mixing ratios of over 1 ppbv, a value that may be unrealistically high since a substantial fraction of HCl expected to have been removed by PSC processing at this time. Following the model results presented in Figure 9, which show HCl increasing fairly steadily throughout the interval from near zero at both levels, the HCl is extrapolated back from the first HALOE data point inside the vortex (on 23 September) by assuming it varies linearly from zero at the start of tilt' study period on 3 September.

In Figure 11 we show the vortex- averaged HCl mixing ratios during the HALOE southern vortex sampling period and the linearl - ext rapolated HCl values during the late winter study period. Also included in Figure 11 are the adjusted mixing ratios of ClO and ClONO₂, the sum ClO+ClONO₂, and the S11111 ClO+ClONO₂ + extrapolated HCl. At both 585 K and 465 K, the sum ClO+ClONO₂ exhibits a decreasing trend over the course of the interval. If ClONO₂ and HCl are the primary reservoirs for chlorine at

this time, then ClO + ClONO₂ + HCl should be roughly a conserved quantity. At 585 K, inclusion of HCl(extrapolated from zero on 3 September) leads to an apparent increase in the chlorine budget of almost ().4 ppbv over the interval, with a final value of ~ 3.0 ppbv. However, the trend in ClO + ClONO₂ + HCl is highly sensitive to the initial value chosen for the HCl extrapolation). Assuming HCl~1 ppbv on 3 September (the result obtained from the $\partial (HCl)/\partial t$ extrapolation) leads to a value of ClO+ClONO₂ + HCl that decreases slightly over the course of the interval, but has a magnitude of about 3.6 ppbv (more than the estimated total chlorine loading at that level). Assuming 11(l=0.5 ppbv on 3 September (a value significantly higher than the model prediction of Figure 9) leads to a fairly constant value of 3.1 ppbv for $ClO + ClONO_2 + HCl$. Thus the chlorin e budget can be made to balance at 585 K. At 465 K, although the addition of extrapolated HCl does improve the chlorine balance slightly, the contribution from calculated HCl is not sufficient to ofl'set the observed losses. The quantity ClO+ClONO₂ + extrapolated HCl declines slightly over the two week study period, although the uncertainties in these quantities are such that a line with zero slope could be fit to these data points. The imbalance in the decrease in reactive chlorine and the increase in the chlorine reservoirs at 465 1 cannot be ameliorated simply by adjusting the extrapolation of the IICl data.

Discussion

The lack of balance between the decrease in ClO and the increase in ClONO₂ in the Antarctic vortex may not be unexpected. Santee et al. [1995] find evidence for severe denitrification in the 1992 Antarctic vortex. Under highly denitrified conditions, the major source of NO₂ is removed and the reformation of ClONO₂ is suppressed. However, if ClONO₂ is not playing a primary role, then the free chlorine must either be going into another reservoir or it must be resupplied. Prather and Jaffe [1990] show that, under conditions of substantial ozone loss, the chemical system is highly nonlinear and the Cl-ClO-ClONO₂ partitioning favors Cl, leading to the preferential formation of HCl.

Douglass et al. [1995] have also recently studied the springtime production of chlorine reservoir species. They use UARS data to constrain a chemical box model, and compare both observations and model results within a narrow annulus just inside the southern vortex boundary (the '(collar)) region) for a 3-month period during the 1992 Antarctic winter to those from a similar region and timespan during the 1991-1992 Arctic winter. Results from model runs at 460 K indicate that, in the absence of denitrification, ozone concentrations dictate which chlorine reservoir dominates in spring. When 03 mixing ratios remain above about 0.5 ppmv, ClONO₂ production arising from HNO₃ photolysis occurs rapidly. However, following the arguments of Prather and Jaffe [1 990], Douglass et al. [1995] find that O₃ mixing ratios below 0.5 ppmv lead to net production of HCl.

The conclusions—of—Douglass et al. [1995] pertain strictly to conditions of mild denitrification (as in the Arctic or the Antarctic collaregion) and extremely low—ozone (as in the Antarctic in October), and consequently—they cannot be applied to the early—to—mid September time period studied—here. While a significant reduction in ozone is occurring throughout September 1992, on the last day of our study period (1.7 September) O₃ concentrations are still above 0.5 ppmv in most of the area of the vortex [Waters et al., 1993b], and the vortex—averaged O₃ mixing—ratio at 465 K [Manney et al., 1993] is still above 1.5 ppmv. However, the formation—of HClmay still be the dominant mechanism for chlorine recovery—at this time. Results from the SLIMCAT is entropic chemical transport 11)0 del[Chipperfield et al., 1995] show that production of ClONO₂ is suppressed and ClO is converted into HCl when the reaction OH + ClO -> HCl + 02—is—included as an 8% channel. Nevertheless, the data seem—to—indicate that the observed decay in enhanced ClO is not entirely offset by a corresponding increase in 110 Tat 465 K.

There are several factors which may account for the apparent lack of balance between the decay of reactive chlorine and the production of chlorine reservoirs at 465 K in the Antarctic vortex. As discussed in connection with Figure 3, there are intermittent localized PSC events at the 465 K level through the end of the study period. Due to the

spacing of the UARS orbit tracks, additional short-lived PSCs with limited geographical extent may have gone undetected. The episodic nature of the PSC activity at the end of southern winter may have created a complicated pattern of recurring chlorine activation and deactivat ion. in fact, vortex- averaged 465 K ClO abundances, while following a general downward trend, are observed to oscillate by a tenth of a ppbv or so on timescales of a few days. It may be that the limited horizontal resolution of the data sets, or the fact that we are examining vortex- averaged quantities at a time when processes are occurring on more localized scales, are obscuring the true behavior of the chlorine species.

Alternatively, the answer may lie inincomplete model chemistry or previously unidentified reservoirs of either active or inactive chlorine. Toonetal. [199'2] estimate the total amount of inorganic chlorine from the HF burden measured using an airborne Fourier Transform Infrared (FTIR) spectrometer during AASE II. They then deduct from this value their measured HCl, ClONO₂, and HOCl burdens, to calculate the amount of "missing" chlorine. Comparing this quantity to an estimated ClO column amount derived from the maximum mixing ratio observed during AASE II plus an equal amount of the ClO dimer, they find that the amount of "missing" chlorine indicated by the FTIR measurements exceeds that likely to reside in ClO and Cl₂O₂. Although this result is obtained for the Arctic vortex, a similar deficit of inorganic chlorine over Antarctica in 1987 is noted by Toon et al. [1989]. Toon et al. [1992] conclude that other unmeasured forms of inorganic chlorine may be present in the winter polar vortex, as suggested by Sander et al. [1 w].

Finally, the quality of the data themselves may be hampering the effort to close the chlorine budget. Difficulties in initializing the chemical transport model [Chipperfield et al., 1 995] reveal that the UARS data set is not entirely self-consistent. For example, the total inorganic chlorine derived from CLAESN₂O measurements using an empirical relationship [Webster et d., 1993] significantly underestimates the sum of MLSCIO and CLAESCIONO₂ at 465–1{. The precision of both the MLS and CLAES data sets is a

limiting factor in the analysis of mixing ratio trends, and it affects our ability to make unequivocal statements about the chlorine budget.

To explore this issue fully requires data extending further into the recovery period. In 1992 the viewing geometry precluded MLS and (.; LAES from making Antarctic measurements during the interval from mid-September through the end of October. By the time solltl)-viewing resumed, ClO was no longer enhanced. Unfortunately, this is the only southern winter for which simultaneous CLAES and MLS data exist; the CLAES supply of cryogen was depleted in May 1993, and there are no CLAES measurements after that time. However, we can use UARS data to investigate the behavior of ClO and the chlorine reservoirs in the northern hemisphere.

1992–1993 Northern Hemisphere Winter

Behavior of C1O and C10NO2

M LS observed lower stratospheric ClO enhanced to greater than 1 ppbv as early as 41 December 1992 [Waters et al., 1 993a; 1 995a], when temperatures briefly fell below the threshold for PSC formation. Temperatures fluctuated around the PSC threshold until late December, after which they were almost continuously below the threshold until late February 1993. When MLS resumed north-viewing measurements in mid-February, lower stratospheric ClOabundances greater than 1 ppbv were seen in most of the vortex region [Waters et d., 1993a]. We will examine this late winter yaw period (12 February 16 March, 1 993) forchlorine balance. As in the southern hemisphere, C LAES measurements of aerosol extinction (not shown) are examined for evidence of continued heterogeneous chlorine activation]. Except for a few transient and highly localized events during the third week in February, no significant PSC activity is observed during the northern hemisphere late winter yaw period, and ClO declines throughout this interval.

Maps of MLS ClO for three days at the beginning, middle, and end of the north-

ern hemisphere late winter yaw period are shown in Figure 12a. Again, only data from the "day" side of the orbit are included in these maps. The northern hemisphere study period is nearly identical to that of the southern hemisphere in terms of season (days since winter solstice); the measurement solar zenith angles are comparable for each set of selected days (cf. Figures 12b and 1b). The same three contours of PV shown in Figure 1a are also superimposed on these maps, but in the northern hemisphere the innermost PV contour shrinks considerably over time and is not suitable for vortex averaging. The PV contours indicate that the vortex is highly distorted and displaced from the pole throughout this period. Although there are patches in which ClO abundances reach Antarctic levels, ClQ is not as uniformly enhanced within the vortex as is observed in the southern hemisphere. In addition, ClO appears to diminish more rapidly at both levels in the northern hemisphere than in the southern hemisphere. At 585 K, chlorine is almost completely deactivated by 1 March. These trends are also evident in Figure 13. Maps of CLAES ClONO $_2$ during the late winter interval are shown in Figure 14. ClONO $_2$ abundances in the Arctic vortex are generally smaller than those in the Antarctic collar region, but there is no pronounced deficit in the vortex interior as there is in the Antarctic (cf. Figure 4). ClONO₂ mixing ratios and the area in the region poleward of 52°S with ClONO₂ concentrations in excess of 1.5 ppbv (Figure 15) show little (or a slightly decreasing) trend at 585 K; however, in contrast to the south, they increase at 465 K.

Effects of Vertical Transport

Diabatic descent is also expected over the northern polar regions [Schoeberl and Hartmann, 1991]. Again, we estimate the vertical velocities at 585 K and 465 K from CLAES measurements of N₂O and CH₄. Daily vortex-averaged (for the intermediate PV contour shown in Figure 12a) mixing ratios for both tracers are plotted in Figure 16. In the north, the vertical velocities estimated from the two tracers are not consistent at 585 K, with CH₄ indicating descent but N₂O indicating ascent. The cause of this

inconsistency may be that the changes in the daily vortex averages cannot be ascribed solely to vertical motions (au underlying assumption in our calculation of the vertical velocities) since two strong stratospheric warmings during the late-winter yaw period [Manney et d., 1 994d] lead to increased horizontal mixing [Manney et d., 1994b]. Based on previous studies [e.g, S. droeberl et al., 1 992], we assume that descent is occurring at this season, and we neglect the N₂O values in intimating the magnitude of the vertical velocity at 585 K. We find the vortex-averaged descent rate to be 0.1 ± 0.2 11111)/s at 5851{ and 0.1 \pm 0.1 mm/s at 465 K when the trends over the entire yaw period are considered. These values are smaller than estimates of the lower stratospheric vertical velocity (0.6-0.7 mm/s) in February 1993 obtained from simulated air parcel trajectories [Manney et al., 19946]. They are also smaller than the late winter residual vertical velocities diagnosed from AASE data by Schoeberl et al. [1992], which range from 0.6 to 1,0 mm/s between 18 and 22 km at 70°N, and the average diabatic descent rates in the lower stratospheric vortex computed from NM(temperatures over a five-month period from November through March by Rosenfield et al. [1 994] (w 1.13-2 km/month, or ~ 0.5 -0.6 mm/s). Contrary to the calculations of Rosenfield et 01. [1 994], our results (and those of Manney et al. [1994b]) indicate that lower stratospheric vertical velocities arc smaller in the northern hemisphere thau in the southern hemisphere (at least for the late winter time periods considered here).

The estimated changes in ClO and ClONO₂ due to vertical transport effects are illustrated in Figure 17. The slight dip apparent in 1 he vortex averaged ClO concentrations just after day 1.5 of the study period (27 February) arises from the fact that measurements obtained in the middle of the yaw cycle have high solar zenith angles. We cannot explain the behavior of the last two ClONO₂ data points (14, 16 March); large data gaps exist on surrounding days (13,15 March), which have 1)('['11 excluded from the vortex averages, but the data coverage on 14 and 16 March is adequate. In general, the inferred Arctic vertical velocities are too weak to have a substantial impact, and the

corrections to the observed mixing ratios are small. The implied chemical loss in ClO is approximately 0.2 ± 0.1 ppbv at 585 K and 1.2 ± 0.1 ppbv at 465 K, based on linear fits to the data. There is a corresponding increase in ClONO₂ of 0.2 ± 0.1 ppbv at 585 K and 1.0 ± 0.1 ppbv at 465 K. In contrast to the southern polar vortex, in the north the calculated chemical changes in ClO and ClONO₂ are balanced (within the uncertainties) at both levels. This result is not highly sensitive to the values of the descent rates; the implied chemical changes in ClO and ClONO₂ are still in balance at both levels, within the limits of the uncertainties, even when vertical velocities as large as those estimated by Schoeberl et al. [1992] are used.

Behavior of HCl

HALOE observations of HCl become available within the northern polar vortex on about 22 March, a few days after MLS and CLAES resume viewing southern high latitudes. HALOE sampling of the northern polar region ceases around 23 April. In the northern hemisphere, sparse data coverage necessitates averaging the HCl within the outermost PV contour shown in Figure 12a (rather than the intermediate PV contour) in order to avoid large gaps in the time series of averaged mixing ratios. The daily vortexaveraged mixing ratios at 585 K and 465 K are presented in Figure 18 for the entire period of northern vortex coverage, along with their associated daily average latitudes. Although HCl exhibits an increasing trend in the weeks after yaw, the increase is much more gradual than that observed in the southern polar region (Figure 10). As in the southern hemisphere, $\partial (HCl)/\partial t$ is calculated through multiple linear regression to account for the latitude dependence of the HALOE data, and used to linearly extrapolate HCl back to the timeframe of the MLS and CLAES data. The extrapolation using the $\partial (HCl)/\partial t$ value leads to HCl mixing ratios on 12 February of about 1.4 ppbv at 585 K and 0.3 ppbv at 465 K, in excellent agreement with the observations from both an aircraft [Webster et al., 1993] and a balloon [Arnold and Spreng, 1994] flight on 13 February 1992 during

AASE II.

In Figure 19 we show the vortex-averaged (over the outermost PV contour) HCl mixing ratios during the HALOE northern vortex sampling period and the linearly-extrapolated HCl values during; the later-winter yaw period. Also included in this figure are the adjusted vortex-averaged (over the intermediate PV contour) mixing ratios of ClOand ClONO2, the S11111 ClO+ClONO2, and the sum ClO+ClONO2+ extrapolated 11(3). At both levels the sum ClO+ClONO2 appears relatively constant (although it undergoes some oscillation at 46,5 K). With the contribution from estimated HCl, the quantity ClO+ClONO2+ HCl is essentially constant at 3.1 ppbv at 585–1<, and varies slightly around \sim 2.5 ppbv at 465–K.

Discussion

Manney et al. [1 994c] show that the 1 992-1 993 Arctic lower stratospheric vortex is anomalously strong, isolated, and persistent, and is characterized by below- average temperatures. It is thus more "Antarctic-like" than is typical. Although the 1992-1993 winter therefore represents fairly extreme Arctic conditions, the observed depiction of gas-phase IINO₃ is nevertheless considerably smaller than that over Antarctica [Santee et al., 1995]. Gas-phase IINO₃ is evidently present in sufficient quantity to supply NO₂ for the rapid formation of ClONO₂. In contrast to the southern hemisphere, we find that the observed late winter decrease in ClO is balanced by a corresponding increase in ClONO₂ at both 585 K and 465 K. This conclusion is in good agreement with the results from the AASE, AASE 11, and EASOE campaigns outlined in the Introduction.

Summary

We have investigated the recovery of the perturbed chlorine chemistry in the lower stratospheric polar regions of both hemispheres. We have correlated MLS measurements of ClO within the polar vortices with simultaneous and co-located CLAES measurements of ClONO₂. Time series of vortex-averaged mixing ratios at both 585 K and 465 K have been analyzed for approximately month-long intervals during late winter: August 17 - September 17, 1 992 in the southern hemisphere and February 11 March 16, 1993 in the northern hemisphere. The magnitude of the diabatic descent at each level has been estimated from ("LAES measurements of CH₄ and N₂O and used to adjust the observed mixing ratios for the effects of vertical transport.

In the southern hemisphere, continuing polar stratospheric cloud activity, as evidenced by CLAES measurements of aerosol extinction coefficient, prevented ClO from undergoing sustained decline until the latter half of the study period. Even after the enhanced (30 abundances started to recede, there was no significant chemical change seen in ClONO₂ at 465 K, and the ClONO₂ at 585 K was observed to decrease. Denitrification is expected to have inhibited the formation of ClONO2 by removing gas-phase HNO3, the source of NO₂. The observational results have also been compared to those from an isent ropic chemical transport model [Chipperfield et al., 1995] which was initialized with UARS data. The model reproduced the MLS ClO abundances and downward trend at 585 K, but underestimated MLS ClO by up to 0.4 ppbv at 465 1{. The model also reproduced CL AES (BONO₂ fairly well at 585 K, but underestimated it by as much as 0.6 ppbv at 465 K. Inclusion of the OH + ClO \rightarrow HCl + O₂ reaction as an 8% channel in the model suppressed the rm'every of ClONO₂ (for the time period considered here) and promoted the conversion of ClO to HCl at both ,5851< and 465 K. 11 A LOE measurements of HCl in the southern polar vortex, obtained shortly after the h'11,S and CLAES data, were linearly extrapolated back to the timeframe of the study period. At 585 K, the trend in ClO+ClONO2 + IICl was found to be highly sensitive to the value to which the IICl was extrapolated on the first day of the interval, and the chlorine budget could be made to balance by setting HCl=0.5 ppby on that day (3 September). At 465 K, although the addition of extrapolated HCl did improve the chlorine balance, the contribution from calculated HCl was not sufficient to offset the observed losses, and a slight imbalance was

found between the decrease in reactive chlorine and the change in the chlorine reservoirs. The difficulty in closing the chlorine budget at 465 K in the southern hemisphere may have arisen from complications caused by ongoing activation, incomplete photochemical assumptions, or inadequate data quality.

In the northern hemisphere, ClO was not as uniformly—enhanced—within the vortex as was observed over—Antarctica. In addition, ClO appeared to diminish more rapidly at both levels in the north ern hemisphere. Although the 1–992-1993 Arctic lower stratospheric vortex was more "Antarctic-like" than is typical, sufficient gas-phase HNO₃ remained to supply NO₂ for the rapid formation of ClO NO₂. In contrast to the southern hemisphere, the decrease in ClO at the end of winter—was—balanced at both 585 K—and 465 K by a corresponding increase in ClONO₂. We found that ClONO₂ was the primary reservoir during recovery in the northern hemisphere, in good agreement with the results from previous Arctic studies.

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This manuscript was prepared with the AGU LATEX macros v3.1.

Figure 1a. Maps of M LS ClO for selected days during the 1992 southern hemisphere late winter south-looking period, interpolated onto the 585 K and 465 K potential temperature surfaces using NM C temperatures. The maps are polar orthographic projections extending to the equator, with the Greenwich meridian at the top and dashed black circles at 30°S and 60°S, and are produced from measurements taken over a 24-hour period. No measurements were obtained in the white area poleward of 80°S. Only data from the "day" side of the orbit are shown. Superimposed in white on each of the maps are three contours of potential vorticity: -0.70 x 10⁻⁴ Km²kg⁻¹s⁻¹, -1.05 x 10⁻⁴ Km²kg⁻¹s⁻¹, and -1.40 x 10" Km²kg⁻¹s⁻¹ at 585 K, and -0.25 x 10⁻⁴ Km²kg⁻¹s⁻¹, -0.375 x 10⁻⁴ Km²kg⁻¹s⁻¹, and -0.50 x 10⁻⁴ Km²kg⁻¹s⁻¹ at 465 1{.

Figure 1b. Maps of the solar zenith angle (sza) of the MLS measurements ("day" side of the orbit) for the days shown in Figure 1a. The 94° sza contour represents the approximate edge of daylight for the measurements.

Figure 2. The area in percent of a hemisphere, as a function of potential temperature and time for the 1992 southern hemisphere late winter south-looking period, within which day time MLS ClO mixing ratios in the region poleward of 52°S exceed 1 ppbv. Tick marks on the vertical axis are not evenly spaced; they represent potential temperature values of 420 K, 465 K, 520 K, 585 I, 655 K, 740 K and 840 K.

Figure 3. The area in percent of a hemisphere as a function of potential temperature and time for the 1992 southern hemisphere late winter south-looking period, within which CLAES aerosol extinction coefficients in the region poleward of 52° S exceed 1.0×10^{-3} km⁻¹. Data gaps (white spaces in plots) occur during the implementation of instrument modes other than normal science operations.

Figure 4. As in Figure 1a (with a different color scale), for CLA ES CIONO₂. Blank spaces in the maps represent areas where there are data gaps or spurious data points.

Figure 5. As in Figure 2 (with a different color scale), for ClONO₂ mixing ratios in excess of 1.5 ppbv. Data gaps (white spaces in plots) occur during the implementation of instrument modes other than normal science operations.

Figure 6. Time series, over the latter half of the 1992 southern hemisphere late winter south-looking period (starting on 3 September), of vortex-averaged (using the intermediate PV contour shown in Figure 1 a - see text) mixing ratios of C LAES CH₄ (ppmv) and N₂O (ppbv) at 585 K and 465 K. The vortex- averaged values for each day are represented by colored circles (black=CH₄, magenta=N₂O), and the lines are the least-squares fits through the data. The slopes of the lines, in combination with the vertical profiles of CH₄ and N₂O, lead to estimates of the diabatic descent rate of 0.5 + 0.3 mm/s at 585 K and 0.74 0.2 mm/s at 465 K.

Figure 7. Vertical profiles of ClO (a) and ClONO₂(b) as a function of potential temperature. These profiles were obtained by averaging individual retrievals both spatially (within the area enclosed by the intermediate PV contour of Figure 1a) and temporally (over 3-17 September).

Figure 8. Time series, over the latter half of the 1992 southern hemisphere late winter south-looking period (starting on 3 September), of vortex-averaged (using the intermediate PV contour shown in Figure 1a) mixing ratios of MLS ClO (ppbv) and CLAES ClONO₂ (ppbv) at 585 K and 465 K. The vortex-averaged values observed on each day are represented by open triangles, and the mixing ratios adjusted for the effects of diabatic descent are given by solid circles.

Figure 9. Vortex- averaged chlorine species at (a) $585\,\mathrm{K}$ and (b) $465\,\mathrm{K}$ from the chemical transport HIOCICI. Solid lines denote model species sampled at $1200\,\mathrm{UT}$; dashed lines denote model species sampled at the same local time as the MLS and CLAES measurements. During the course of the day, the vortex moves relative to the PV contour (fixed at $1200\,\mathrm{UT}$) used here to delimit the area over which vortex averages are taken. This occasionally causes model ClO sampled at the MLS local time to exceed model ClO_x sampled at $1200\,\mathrm{UT}$. The curves representing model ClO sampled at $1200\,\mathrm{UT}$ (not shown) and model ClO_x sampled at $1200\,\mathrm{UT}$ do not cross. Also shown are the MLS ClO and CLAES (ClONO₂ values. These differ slightly from the vortex averages shown in Figure 8 because they are calculated from the ClO and ClONO₂ fields interpolated onto the model T42 Gaussian grid.

Figure 10. Time series, over the HALOE1992 southern vortex sampling period, of vortex-averaged (within the intermediate PV contour) mixing ratios of HALOEHC1(ppbv) at (a) 585 K and (b) 465 K. The associated daily average latitudes of the profiles are shown in (c) (open squares = 465 K, solid circles = 585 K). HALOE data are strong functions of both latitude and time, and multiple linear regression is used to fit curves (solid lines) to the daily averages at both levels.

Figure 11. Vortex- averagedHCl mixing ratios during, the HALOE 1 992 southern vortex sampling period (green circles); linearly - extrapolated 11(:1 values (see text) during the latter half Of the late-winter yaw period (dotted green line); adjusted vortex averaged mixing ratios of MLS ClO (red circles) and CLAES ClONO₂ (cyan circles); the sum (lO + ClONO₂ (black circles); and the quantity ClO + ClONO₂ + extrapolated HCl (magenta pluses).

Figure 12a. As in Figure 10, for the 1992-1993 northernhemisphere late winternorth-looking period, with the Greenwich meridian at the bottom, and positive PV contour values.

Figure 12b. As in Figure 1b, for the days shown in Figure 12a.

Figure 13. As in Figure 2 (with a different color scale), for the 1992-1993 northern hemisphere late winter north-looking period.

Figure 14. As in Figure 12a, for CLAES ClONO₂. Blank spaces in the maps represent areas where three are data gaps or spurious data points.

Figure 15. As in Figure 13, for ClONO₂ mixing ratios in excess of 1.5 ppbv. Data gaps (white spaces in plots) occur during the implementation of instrument modes other than normal science operations.

Figure 16. As in Figure 6, for the 1992-1993 northern hemisphere late winter north--looking period. The slopes of the lines lead to estimates of the diabatic vertical velocity of $0.1 \pm 0.2 \,\mathrm{mm/s}$ at $585 \,\mathrm{K}$ and $0.1 \pm 0.1 \,\mathrm{mm/s}$ at $465 \,\mathrm{K}$.

Figure 17. As in Figure 8, for the 1992-1993 northern hemisphere late winternorth-looking period.

Figure 18. As in Figure 10, for tile HALOE 1992-199:3 northern vortex sampling period. In this case, averages are computed over the entire vortex region (see text).

Figure 19. As in Figure 11, for the northern hemisphere.

MLS C10

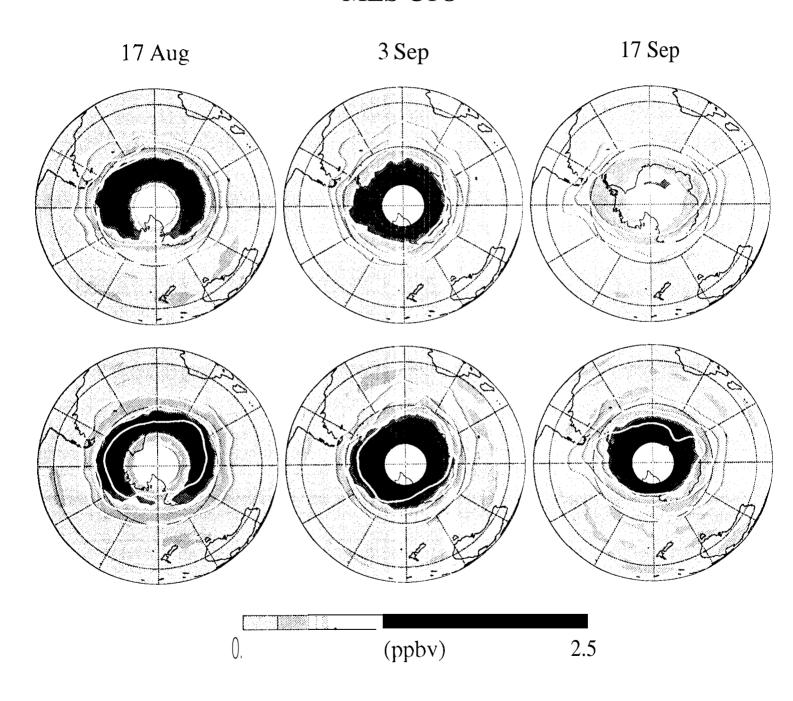


Figure 1a

Solar Zenith Angle

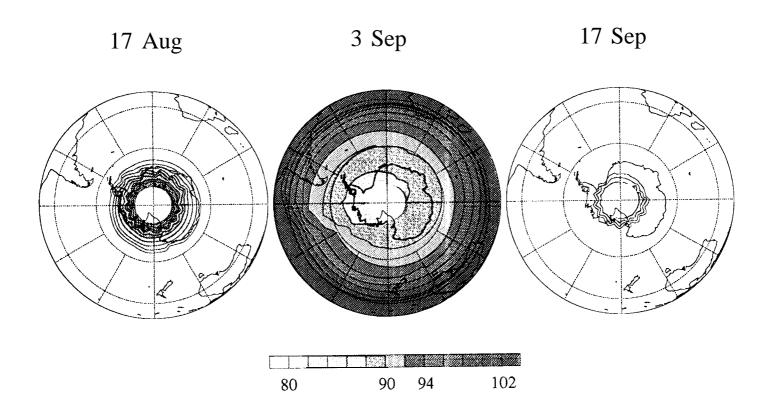
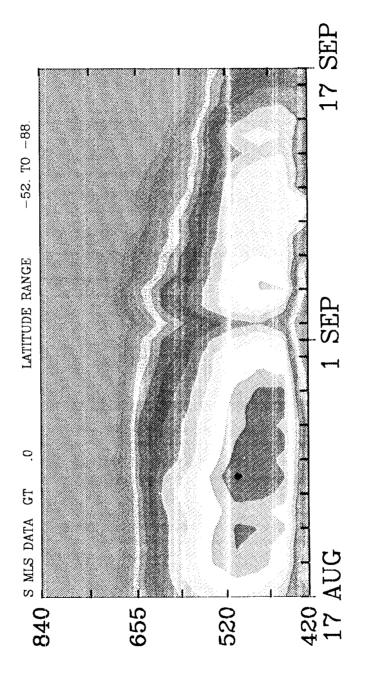
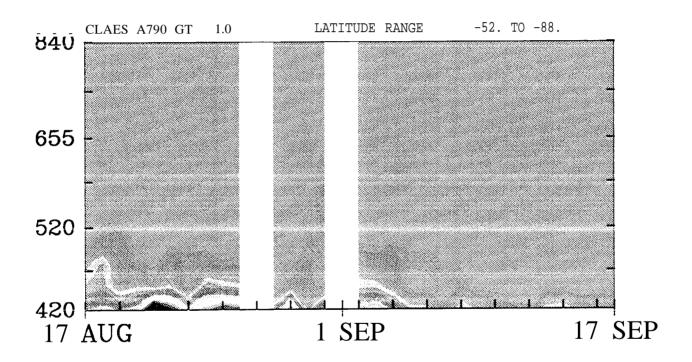


Figure 26

0.0300





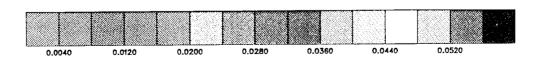


Figure 3

CLAES C10NO₂

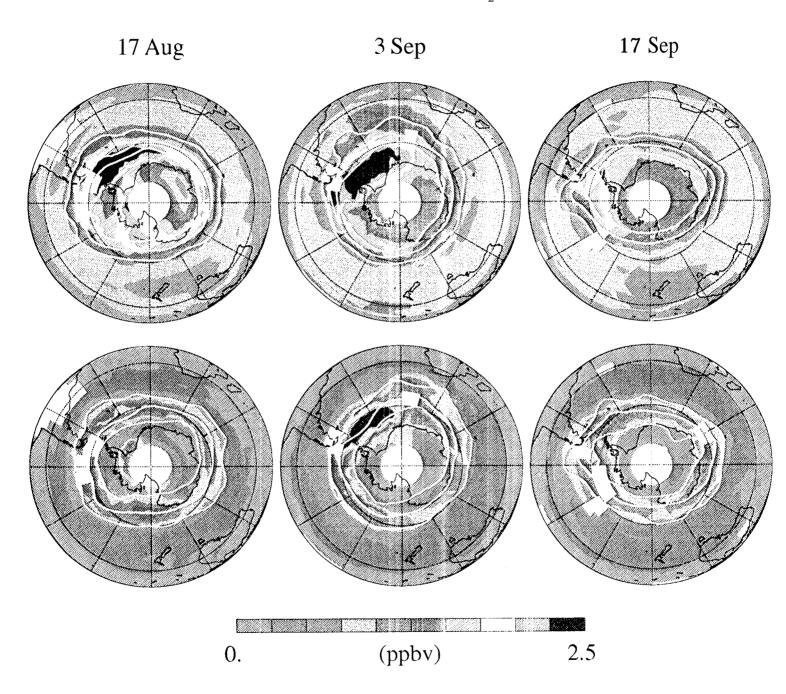
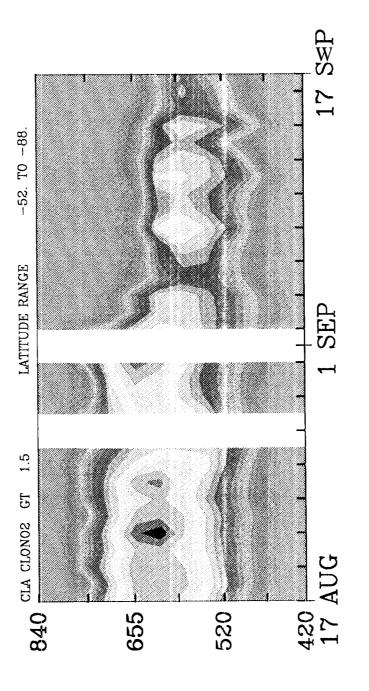


Figure 4



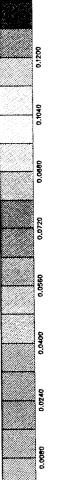
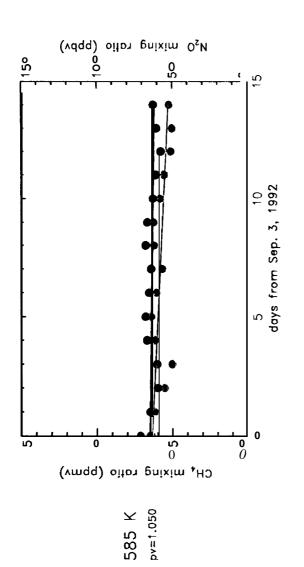
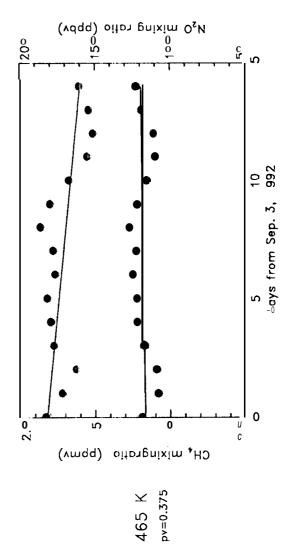
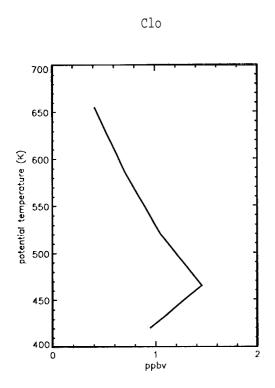


Figure 5







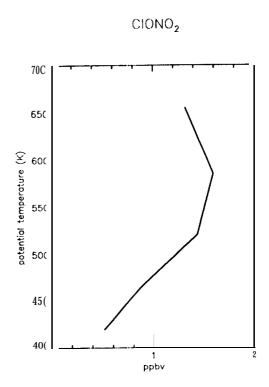


FIGURE 70,b

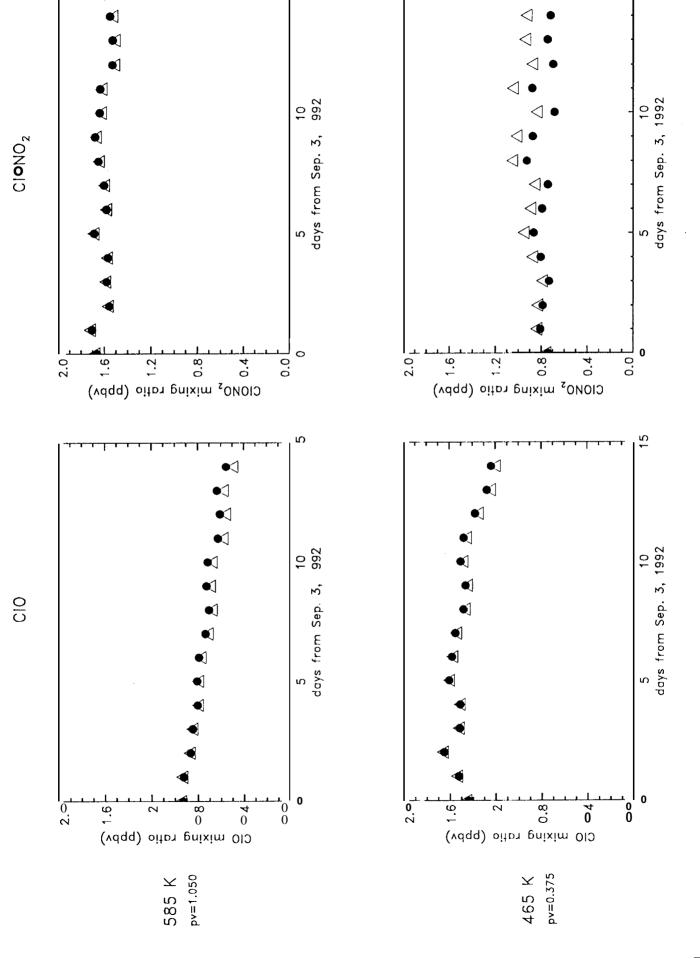


FIGURE 8

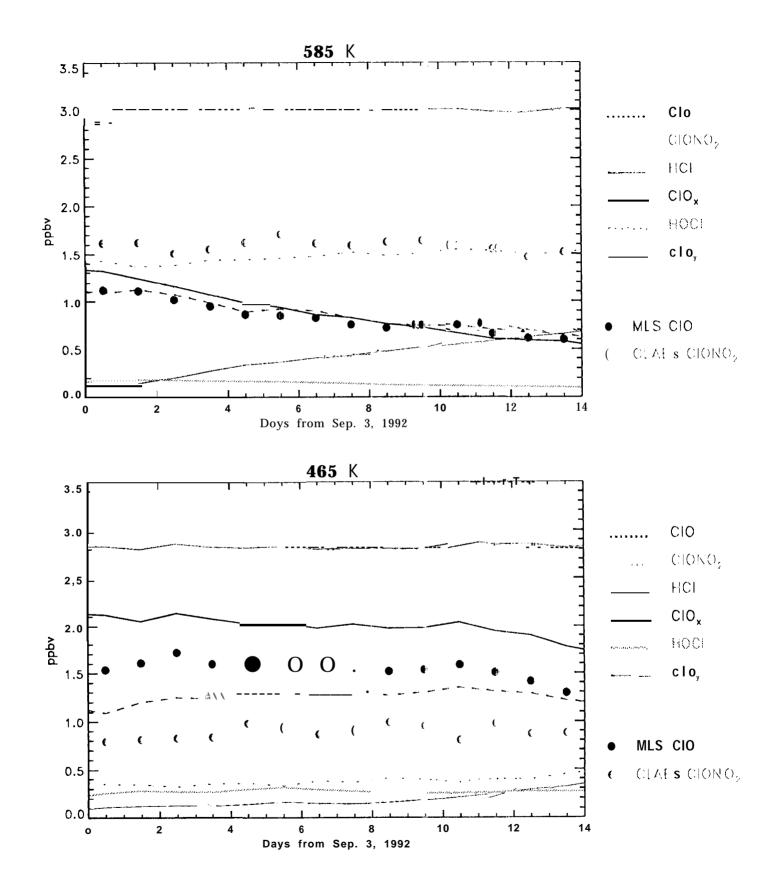


Figure 9

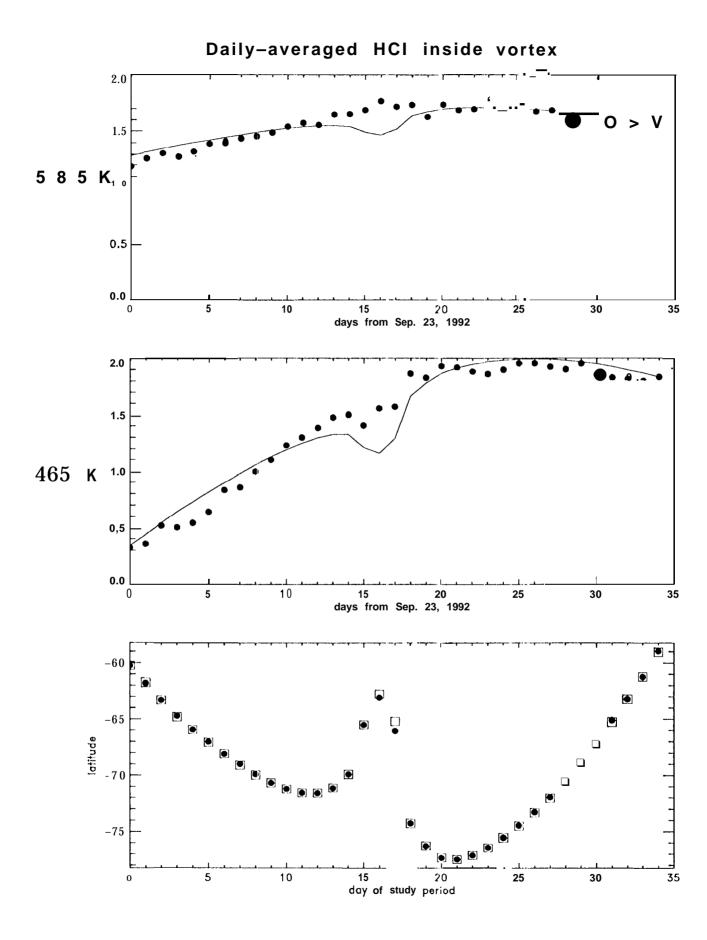


Figure 10

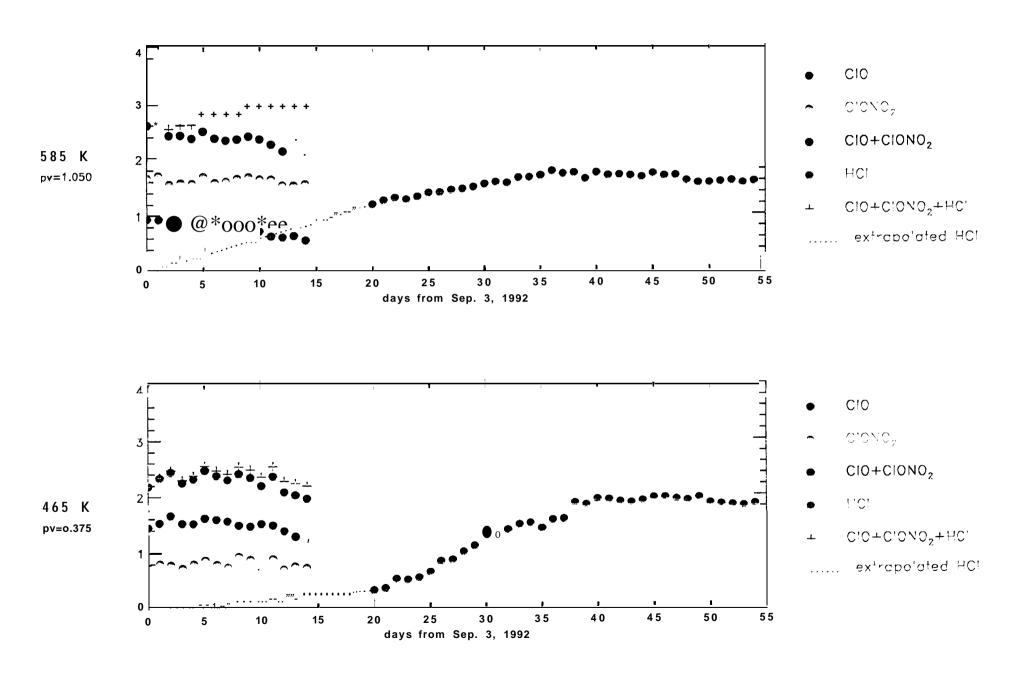


Figure 11

MLS C10

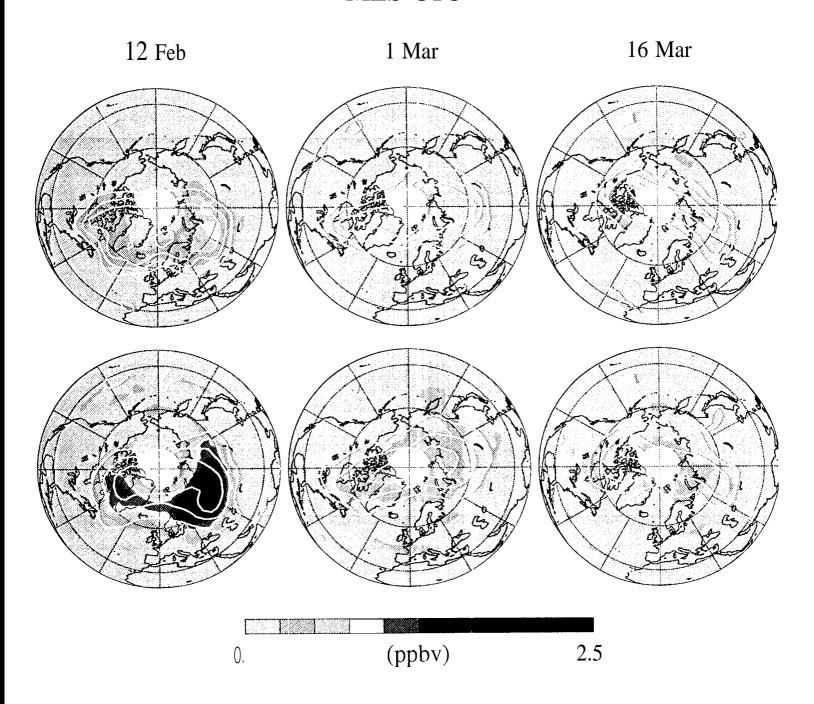


Figure 12a

Solar Zenith Angle

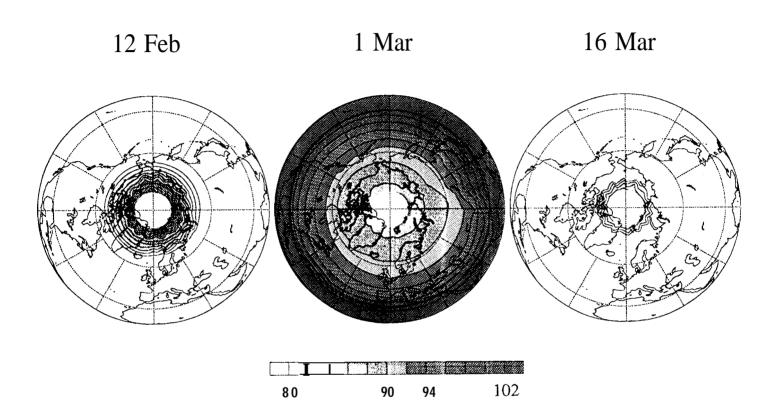
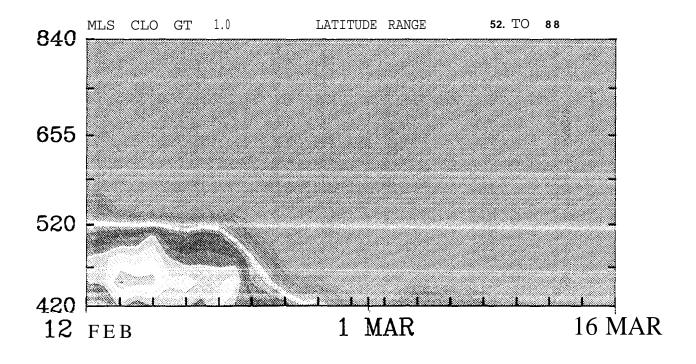


Figure 126



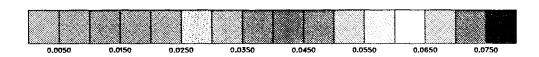


Figure 13

CLAES CIONO₂

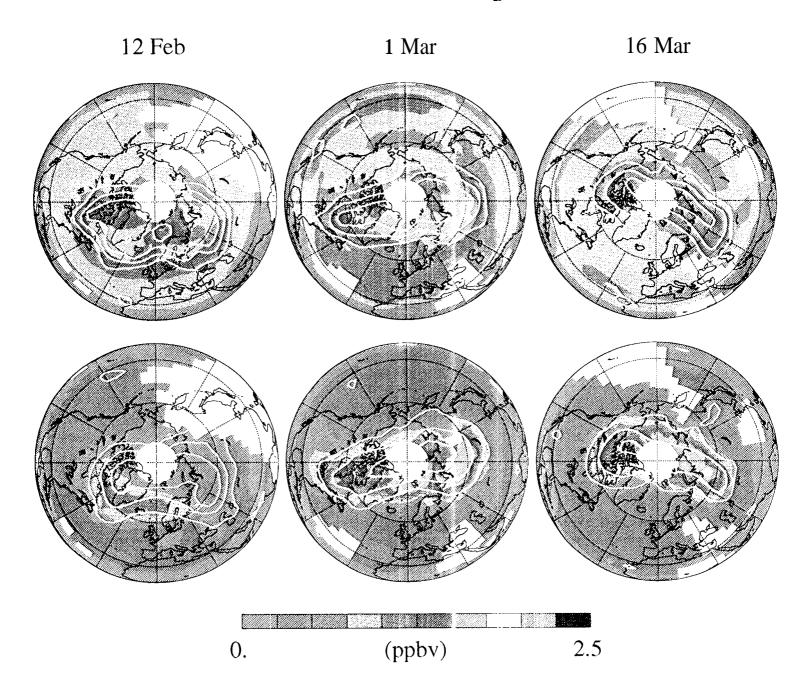
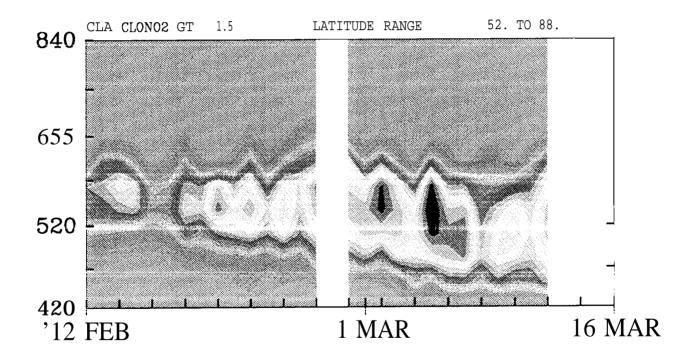


Figure 14



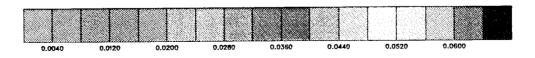
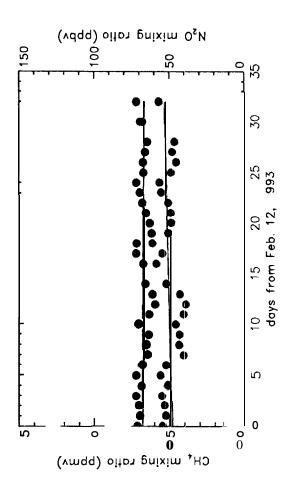
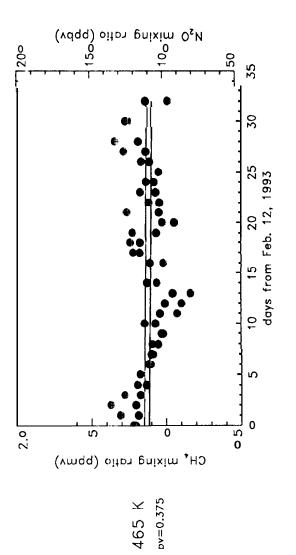


Figure 15



585 K pv=1.050



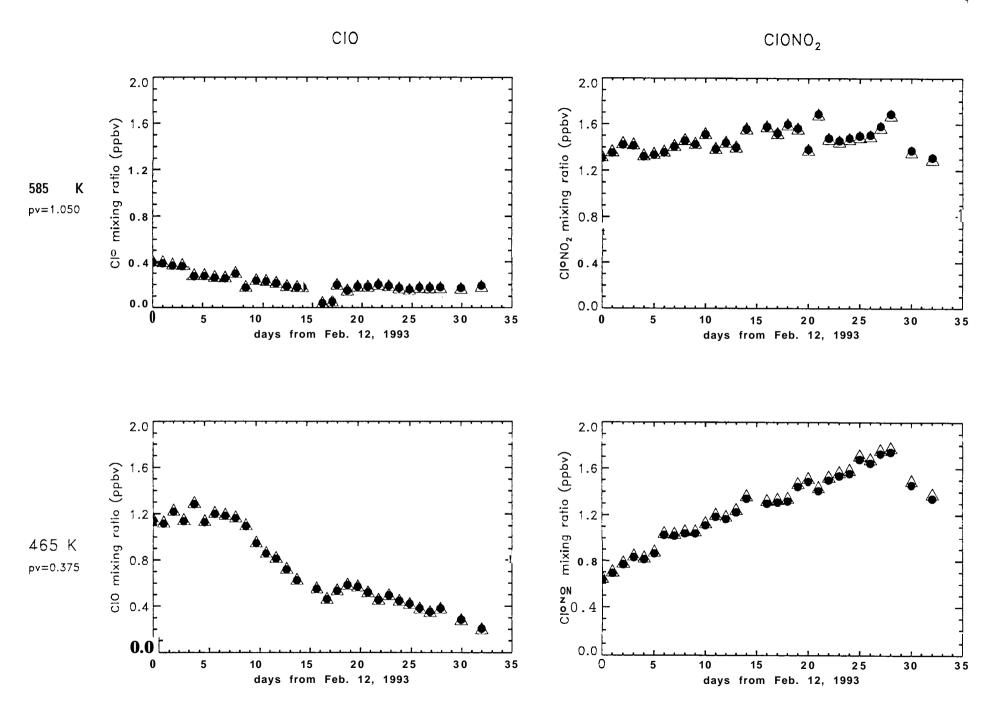


FIGURE 17

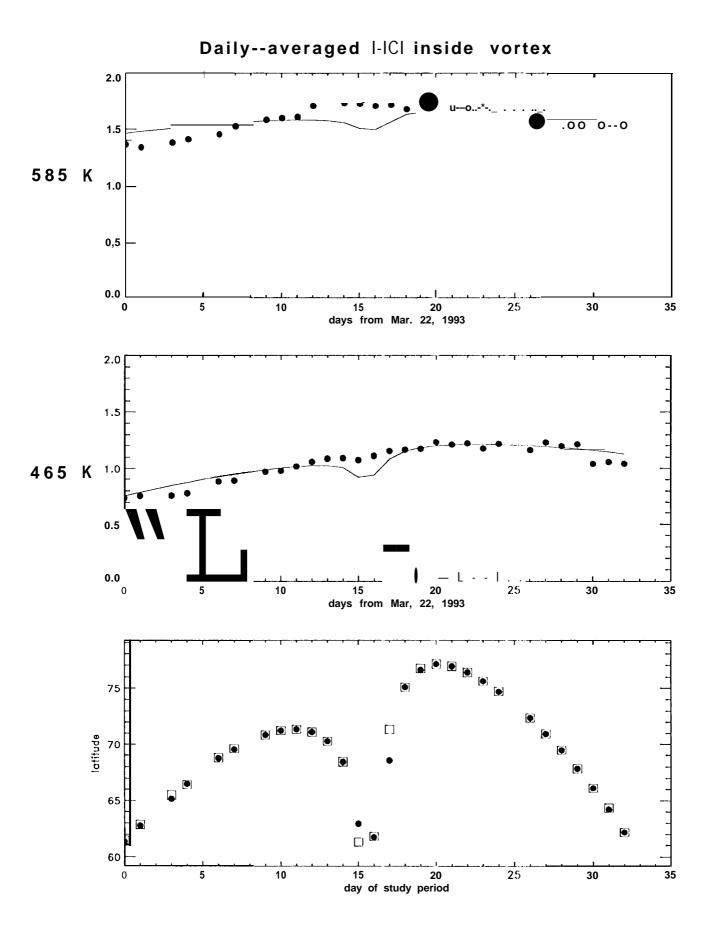


Figure 18

